Hydroclassified Combinatorial Saturation Mutagenesis: Reshaping Substrate Binding Pockets of KpADH for Enantioselective Reduction of Bulky–Bulky Ketones

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Supporting Information

ABSTRACT: A hydroclassified combinatorial saturation mutagenesis (HCSM) strategy was proposed for reshaping the substrate binding pocket by dividing 20 amino acids into four groups based on their hydrophobicity and size. These smart HCSM libraries could significantly reduce screening effort especially for the simultaneous mutagenesis of three or more residues and lacking high throughput screening methods. Employing HCSM strategy, the stereoselectivity of KpADH, an alcohol dehydrogenase from Kluyveromyces polysporus, was efficiently improved to 99.4% ee. (4-Chlorophenyl)(pyridin-2-yl)methanone (CPMK), generally regarded as a "hard-to-reduce" ketone, was used as a model substrate, and its corresponding chiral alcohol products could be utilized as antihistamine precursors. The best variant S0C10 displayed higher binding affinity and catalytic efficiency toward CPMK with $K_M/k_{cat}$ of 59.3 s$^{-1}$mM$^{-1}$, 3.51-fold that of KpADH. Based on MD simulations, increased difference between two binding pockets, enhanced hydrophobicity, and $\pi-\pi$ and halogen–alkyl interactions were proposed to favor the enantioselective recognition and substrate binding in S0C10. Substrate spectrum analysis revealed that S0C10 exhibited improved enantioselectivity toward diaryl ketones especially with halo- or other electron-withdrawing groups. As much as 500 mM CPMK could be asymmetrically reduced into chiral diaryl alcohols with ee of 99.4% and a space–time yield of 194 g L$^{-1}$d$^{-1}$ without addition of external NADP$^+$. This study provides an effective mutagenesis strategy for the protein engineering of substrate specificity and enantioselectivity.

KEYWORDS: alcohol dehydrogenase, hydroclassified amino acids, directed evolution, enantioselectivity, (R)-CPMA

INTRODUCTION

Optically active diarylmethanols are important building blocks and structural motifs of pharmaceuticals, such as antihistamine, diuretic, antiepileptic, asthma, and antidepressant drugs (Scheme 1).1 Moreover, via S$_2$N2 substitution at the C–O bond, chiral diarylmethanols could be facilely converted into diarylmethane derivatives without loss of optical purity.2 As a result, efficient synthesis of chiral diarylmethanols, typically through asymmetric addition of aryl nucleophiles benzaldehyde or asymmetric transfer hydrogenation (ATH) of the corresponding diaryl ketones, is of special interest.3

Compared with ATH using precious metals, such as iridium, rhodium, and ruthenium as catalysts, carbonyl reductases mediated reactions have many inherent merits such as environmental compatibility, high enantioselectivity, and mild reaction conditions.4 Thus, asymmetric bioreduction has attracted ever-increasing attention, and various reductases/dehydrogenases have been validated at large scale.5 However, the enantioselective reduction of bulky–bulky diaryl ketones

with two similar aryl substituents remains challenging for not only chemocatalysts but also biocatalysts, especially at high substrate loading.6 Only SSCR from Sporabolomyces salmonicolor and commercial ketoreductase KRED124 have been reported with 88–94% ee at 10 g L$^{-1}$ (4-chlorophenyl)-(pyridin-2-yl)methanone (CPMK, 1a) in depletion of 1 g L$^{-1}$ NADP$^+$.7 Previously, KpADH has been identified from Kluyveromyces polysporus by genome database mining with 82.5% (R) ee in the reduction of 100 mM CPMK.8 The KpADH displays the highest $k_{cat}/K_M$ among all the reported reductases, indicating it is a promising candidate for further protein engineering. Directed evolution is a commonly practiced approach for protein engineering as an alternative to rational design.9 Site-directed saturation mutagenesis is a major strategy performed

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at sites lining the binding pockets of enzymes to improve their limited substrate scope, insufficient activity, and poor regio- or stereoselectivity.¹⁰ Although numerous successes in obtaining enzymes with improved catalytic properties have been achieved by directed evolution, many uncertainties still persist to be clarified, for example, how to group a variety of residues into randomization site, how to choose optimal upward pathways, etc.¹¹ As the number of residues in a randomization site increases to five or more, the screening effort for 95% library coverage, as calculated by CASTER based on the Patrick/Firth algorithm, is astronomically high.¹² Therefore, the construction of libraries for multiple sites with reasonable size is still challenging, especially in the engineering of certain properties, such as stereoselectivity, for which a high-throughput screening method is not available.¹³ To solve the labor-intensive bottleneck, three tricks have been proposed to create “smart” mutant libraries: (1) binary patterning method employing polar/nonpolar residues, (2) splitting up a large number of sites into smaller ones followed by iterative saturation mutagenesis, and (3) use of reduced amino acid alphabets.¹⁴ NNK (32:20, codons:amino acids), NDT (12:12), or 22c-trick (22:20) degeneracy have been commonly adopted in directed evolution to search the whole protein sequence space.¹⁵ However, the concurrent oversampling increases as the residues in randomization site increase from one residue to more, and the inevitable codon bias emerges with the redundant degenerated codons. In the simultaneous mutation of three residues as illustrated in Scheme 2, the coverage rates of 2000−3000 variants (generally accepted screening work for enantioselectivity engineering) are 14% and 38% for NNK and

Scheme 1. Stereoview of the Homology Structure of KpADH and Asymmetric Reduction of CPMK into (R)-CPMA Catalyzed by KpADH. (A) Homology Structure of KpADH; (B) Asymmetric Reduction of CPMK into (R)-CPMA; (C) Residues Lining the Substrate Binding Pockets*¹⁴

Scheme 2. Hydroclassified Combinatorial Saturation Mutagenesis Alphabets and the Screening Effort Required for Combinatorial Saturation Mutagenesis of Two or Three Residues Employing NNK/S, 22c-Trick, and HCSM Strategies*¹⁴

“Dashed lines, 2 residues; solid lines, 3 residues; red lines, NNK/S; purple lines, 22c-trick; green lines, HCSM. Number of variants is calculated according to the formula \( L = -V \times \ln(1 - F) \), where \( L \) is the number of variants or library size, \( V \) is the theoretical number of variants, and \( F \) is the expected coverage of library, e.g., 0.95 for 95% coverage.

“Catalytic triad illustrated in green stick, NADPH in yellow stick, CPMK in magenta stick, residues of substrate binding pockets in cyan ball and stick.
RESULTS AND DISCUSSION

Alanine and Tyrosine Scanning for Hotspots. A proposed mechanism for asymmetric carbonyl reduction catalyzed by short chain dehydrogenase/reductase (SDR) has been well reviewed with Ser-Tyr-Lys as catalytic triad.\(^\text{18}\) Catalytic residues Ser and Tyr are responsible for the stabilization of alkoxide formed on hydride transfer from NADPH and nucleophilic attack, while Lys is essential for the stabilization of substrate and activation of water. It has been recognized that there are small and large binding pockets (Poc\(_S\) and Poc\(_L\)) in SDR which could discriminate and accommodate prochiral ketones with substituents differing in size.\(^\text{19}\) Physical and chemical properties of Poc\(_S\) and Poc\(_L\), determined by their residues constitution, are vital for catalytic properties, especially substrate specificity and enantioselectivity. A homology model of KpADH was constructed employing crystal structures of methylglyoxal/isovaleraldehyde reductase from Saccharomyces cerevisiae (51% identity, PDB: 4PVC) and ketoreductase 1 from Candida glabrata (50% identity, PDB: 5B6K) as templates.\(^\text{20}\) Evaluation of the homology model of Candida glabrata ketoreductase 1 from especially substrate specificity, are vital for catalytic properties, Poc\(_S\) or Poc\(_L\), determined by their residues constitution, are vital for catalytic properties, especially substrate specificity and enantioselectivity. A homology model of KpADH was constructed employing crystal structures of methylglyoxal/isovaleraldehyde reductase from Saccharomyces cerevisiae (51% identity, PDB: 4PVC) and ketoreductase 1 from Candida glabrata (50% identity, PDB: 5B6K) as templates.\(^\text{20}\) Evaluation of the homology model of KpADH reveals that over 99.4% of residues are located in the allowable regions in the Ramachandran plot (Figure S1). The proposed substrate binding pockets of KpADH are made up of 16 residues according to its homology model, including V84, F86, Y127, A128, M131, P133, Q136, F161, C165, S196, F197, E214, S237, T234, Q238, and I308 (Table 1). Simultaneous saturation mutagenesis of all the residues in substrate binding pockets is obviously beyond reach. Moreover, most of the above 16 residues locate in the loop region and might not be correctly assigned into Poc\(_S\) or Poc\(_L\) according to the homology model. As a result, we attempted to identify the hotspots by alanine and tyrosine scanning and then classify them into Poc\(_S\) or Poc\(_L\), which were presumed to accommodate 2’-pyridyl and 4’-chlorophenyl groups of CPMK, respectively. Mutation of Poc\(_S\) residues into larger amino acids might cause a reduction in the size of Poc\(_S\) and thus increased volumetric difference between Poc\(_S\) and Poc\(_L\), whereas mutation into smaller amino acids could lead to enlarged Poc\(_S\) and decreased difference between two pockets. Presumably, variants with mutations at Poc\(_S\) residues into larger amino acids might exhibit higher enantioselectivity than mutations into smaller amino acids. Likewise, variants with mutation at Poc\(_L\) residues into smaller amino acids might exhibit higher enantioselectivity than mutations into larger amino acids.

Preliminary screening was therefore performed by mutation of each residue into alanine and tyrosine (namely alanine and tyrosine scanning method),\(^\text{21}\) due to their moderate hydrophobicity (+1.8 for alanine and −1.3 for tyrosine), different size, and additional hydroxyl group of tyrosine. Scanning results revealed that 11 residues participate in influencing the enantioselectivity of KpADH toward CPMK, while the other five residues V84, F86, A128, P133, and T234 were eliminated due to their unstable ee of 15−83% (V84, data not shown) or similar ee values compared with KpADH (Table 1). As shown in Table 1, variants with mutations at residues Y127, Q136, F161, and S196 exhibited similar enantioselectivity between alanine and tyrosine mutations. As a result, they were not selected as hotspots, in spite of their different enantioselectivity in comparison with KpADH. The ee values of I308A and I308Y were 76.8% and 73.4% respectively, lower than 82.5% ee of KpADH. Thus, site 308 was also not chosen as a hotspot. Mutations at M131, C165, F197, E214, S237, and Q238 resulted in highly improved ee. Hence, these six sites were selected as mutational hotspots. The alanine mutants M131A (88.6%), F197A (94.2%), and S237A (96.1%) exhibited higher ee values than their corresponding tyrosine mutants M131Y (86.7%), F197Y (86.6%), and S237Y (94.1%), respectively. The results suggest that M131, F197, and S237 could be key sites in Poc\(_S\) due to their enhanced stereoselectivity when mutated into smaller amino acids. The alanine mutants C165A (75.5%), E214A (79.8%), and Q238A (79.3%) displayed lower ee values than their corresponding tyrosine mutants C165Y (88.1%), E214Y (93.8%), and Q238Y (87.7%), respectively. It indicates that C165, E214, and Q238 could be key sites in Poc\(_S\) since higher ee values were achieved when mutated into larger amino acids. Consequently, six hotspots were identified to play critical roles in manipulating the enantioselectivity of KpADH, and they were grouped into Poc\(_S\) and Poc\(_L\) residues for further study.

<table>
<thead>
<tr>
<th>residue</th>
<th>ee of alanine mutation (%)</th>
<th>ee of tyrosine mutation (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>F86</td>
<td>82.7 ± 0.3</td>
<td>75.6 ± 0.5</td>
</tr>
<tr>
<td>Y127</td>
<td>78.7 ± 0.6</td>
<td>82.5 ± 0.6 (77.9 ± 0.5)</td>
</tr>
<tr>
<td>A128</td>
<td>82.5 ± 0.6</td>
<td>82.9 ± 0.3 (82.8 ± 0.2)</td>
</tr>
<tr>
<td>M131</td>
<td>88.6 ± 0.3</td>
<td>86.7 ± 0.5</td>
</tr>
<tr>
<td>P133</td>
<td>85.3 ± 0.7</td>
<td>82.7 ± 0.3</td>
</tr>
<tr>
<td>Q136</td>
<td>78.3 ± 0.8</td>
<td>79.7 ± 0.3</td>
</tr>
<tr>
<td>F161</td>
<td>67.6 ± 0.5</td>
<td>67.5 ± 0.3</td>
</tr>
<tr>
<td>C165</td>
<td>75.5 ± 0.3</td>
<td>88.1 ± 0.5</td>
</tr>
<tr>
<td>S196</td>
<td>74.7 ± 0.5</td>
<td>74.7 ± 0.3</td>
</tr>
<tr>
<td>F197</td>
<td>94.2 ± 0.5</td>
<td>86.6 ± 0.5</td>
</tr>
<tr>
<td>E214</td>
<td>79.8 ± 0.3</td>
<td>93.8 ± 0.2</td>
</tr>
<tr>
<td>S237</td>
<td>96.1 ± 0.6</td>
<td>94.1 ± 0.4</td>
</tr>
<tr>
<td>Q238</td>
<td>79.3 ± 0.2</td>
<td>87.7 ± 0.5</td>
</tr>
<tr>
<td>T234</td>
<td>82.1 ± 0.5</td>
<td>83.7 ± 0.3</td>
</tr>
<tr>
<td>I308</td>
<td>76.8 ± 0.3</td>
<td>73.4 ± 0.4</td>
</tr>
</tbody>
</table>

*Number in the parentheses denotes the ee of mutants with phenylalanine mutation.

Development and Screening of HCSM Libraries. Considering that the enantioselectivity could be manipulated...
by more than one residue in substrate binding pockets, combinatorial mutagenesis is often performed. To conduct the combinatorial saturation mutagenesis with high efficacy, hydroclassified amino acid alphabet was proposed by classifying the codons into four groups based on the codon bias of *E. coli*, hydrophobicity, and size of amino acids. As illustrated in Scheme 2, 24 codons were chosen for 20 amino acids. As illustrated in Scheme 2, 24 codons were chosen for 20 amino acids and were classified into Group I (designated as "Hydrophobic group", VYG), Group II ("Hydrophilic group", VRT), Group III ("Mixed group", VWA), and Group IV ("Aromatic group", TDS). Specifically, Group I codes for threonine, methionine, alanine, valine, proline, and leucine. Group II codes for aspartate, asparagine, serine, glycine, histidine, and arginine. Group III codes for lysine, isoleucine, glutamate, valine, glutamine, and leucine. Group IV codes for tyrosine, tryptophan, phenylalanine, cysteine, leucine, and stop codons. Saturation mutagenesis employing this hydroclassified amino acid alphabet is named as hydroclassified combinatorial saturation mutagenesis (HCSM). Compared with other combinatorial mutagenesis strategies for engineering enantioselectivity such as triple-codon saturation mutagenesis (TCSM), double-codon saturation mutagenesis (DCSM), and single-codon saturation mutagenesis (SCSM), HCSM is a hexatruple-codon saturation mutagenesis method in which each site is mutated into six other amino acids. It is therefore supposed to be more efficient since more codons are employed in randomizing the mutation sites (TCSM > DCSM > SCSM).

Most importantly, in HCSM, all hotspots could be simultaneously mutated into hydrophobic (Group I), hydrophilic (Group II), or aromatic (Group IV) amino acids, as well as the mixed hydrophobic and hydrophilic codons in Group III, which could conduce to the evaluation of possible synergistic effects of hydrophobicity and hydrophilicity. Consequently, HCSM is advantageous for developing libraries with high diversity in volume, hydrophobicity, substituents, and electricity.

**Results of HCSM Libraries in PocL.** Furthermore, the HCSM libraries of KpADH at the above six hotspots in PocL and PocL were constructed. For M131, F197, and S237 in PocL, all three sites were combinatorial saturation mutated adopting the

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**Figure 1.** Screening results of HCSM libraries on large and small substrate binding pockets. (A) PocL-I, (B) PocL-II, (C) PocL-III, (D) PocL-IV, (E) PocS-I, (F) PocS-II, (G) PocS-III, (H) PocS-IV, and (I) combinatorial mutagenesis.
Table 2. Kinetic Parameters of KpADH and Variants toward CPMK

<table>
<thead>
<tr>
<th>variant</th>
<th>library</th>
<th>mutation</th>
<th>ee (%)</th>
<th>$K_M$ (mM)</th>
<th>$k_{cat}$ (s$^{-1}$)</th>
<th>$k_{cat}/K_M$ (s$^{-1}$·mM$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>KpADH</td>
<td>–</td>
<td>–</td>
<td>82.5 ± 0.5</td>
<td>0.78 ± 0.07</td>
<td>13.2 ± 0.3</td>
<td>16.9</td>
</tr>
<tr>
<td>3D3</td>
<td>PocC-1</td>
<td>M131A/F197T/S237A</td>
<td>98.2 ± 0.3</td>
<td>3.84 ± 0.11</td>
<td>8.23 ± 0.72</td>
<td>2.14</td>
</tr>
<tr>
<td>3H8</td>
<td>PocC-1</td>
<td>M131A/S237A</td>
<td>97.8 ± 0.4</td>
<td>1.95 ± 0.10</td>
<td>21.3 ± 0.4</td>
<td>10.9</td>
</tr>
<tr>
<td>6C12</td>
<td>PocC-1</td>
<td>M313/F197T/S237A</td>
<td>98.5 ± 0.3</td>
<td>2.75 ± 0.15</td>
<td>12.7 ± 0.5</td>
<td>4.62</td>
</tr>
<tr>
<td>43D11</td>
<td>PocC-IV</td>
<td>C165F/E214Y/Q238C</td>
<td>97.4 ± 0.5</td>
<td>1.51 ± 0.08</td>
<td>15.8 ± 0.4</td>
<td>10.46</td>
</tr>
<tr>
<td>45A3</td>
<td>PocC-IV</td>
<td>C165F/E214Y</td>
<td>97.5 ± 0.3</td>
<td>0.68 ± 0.05</td>
<td>14.9 ± 0.5</td>
<td>21.91</td>
</tr>
<tr>
<td>47G9</td>
<td>PocC-IV</td>
<td>C165F/E214Y/Q238C</td>
<td>97.7 ± 0.5</td>
<td>1.46 ± 0.09</td>
<td>13.2 ± 0.4</td>
<td>9.04</td>
</tr>
<tr>
<td>49A7</td>
<td>Comb. Lib.$^b$</td>
<td>M131V/F197T/E214Y/S237A</td>
<td>98.9 ± 0.2</td>
<td>1.88 ± 0.07</td>
<td>28.4 ± 0.5</td>
<td>15.1</td>
</tr>
<tr>
<td>49D6</td>
<td>Comb. Lib.</td>
<td>C165F/E214Y/S237A/Q238C</td>
<td>98.9 ± 0.3</td>
<td>2.62 ± 0.08</td>
<td>22.0 ± 0.3</td>
<td>8.40</td>
</tr>
<tr>
<td>50C10</td>
<td>Comb. Lib.</td>
<td>C165F/E214Y/S237A</td>
<td>99.4 ± 0.2</td>
<td>0.35 ± 0.05</td>
<td>20.8 ± 0.3</td>
<td>59.4</td>
</tr>
<tr>
<td>51E10</td>
<td>Comb. Lib.</td>
<td>M131A/C165F/F197T/E214Y/S237A</td>
<td>98.7 ± 0.4</td>
<td>6.52 ± 0.13</td>
<td>55.4 ± 1.2</td>
<td>8.50</td>
</tr>
<tr>
<td>M131V</td>
<td>SDM$^c$</td>
<td>M131V</td>
<td>87.4 ± 0.6</td>
<td>1.14 ± 0.07</td>
<td>9.39 ± 0.21</td>
<td>8.24</td>
</tr>
<tr>
<td>C165F</td>
<td>SDM</td>
<td>C165F</td>
<td>89.1 ± 0.3</td>
<td>0.52 ± 0.04</td>
<td>9.26 ± 0.22</td>
<td>17.8</td>
</tr>
<tr>
<td>F197T</td>
<td>SDM</td>
<td>F197T</td>
<td>87.4 ± 0.5</td>
<td>8.12 ± 0.13</td>
<td>6.80 ± 0.19</td>
<td>0.84</td>
</tr>
<tr>
<td>E214Y</td>
<td>SDM</td>
<td>E214Y</td>
<td>94.2 ± 0.3</td>
<td>0.80 ± 0.06</td>
<td>14.9 ± 0.6</td>
<td>18.6</td>
</tr>
<tr>
<td>S237A</td>
<td>SDM</td>
<td>S237A</td>
<td>96.1 ± 0.3</td>
<td>0.38 ± 0.04</td>
<td>15.4 ± 0.7</td>
<td>40.5</td>
</tr>
<tr>
<td>Q238C</td>
<td>SDM</td>
<td>Q238C</td>
<td>88.4 ± 0.3</td>
<td>3.42 ± 0.14</td>
<td>10.2 ± 0.3</td>
<td>2.98</td>
</tr>
</tbody>
</table>

$^a$Comb. Lib.: combinatorial mutagenesis library. $^b$SDM: site-directed mutagenesis.

hydroclassified amino acid alphabet (Groups I, II, III, and IV) to generate HCSM libraries of PocC-1, PocC-2, PocC-3, and PocC-4, respectively. Theoretically, at least 647 variants are required for each library to achieve 95% coverage according to the formula in Scheme 2. In total, at least 2588 variants are required for four libraries of PocC. This library size is still too large for enantioselectivity analysis by chiral HPLC. As a result, two rounds of screening were adopted to isolate KpADH variants with both high activity and enantioselectivity. First, deactivated variants were eliminated by high throughput screening on activity. Then, biotransformation employing the active variants was conducted for enantioselectivity analysis. As shown in Figure 1B, 98% of the variants of PocC-2 were deactivated, and only 15 variants displayed 82–94% ee and less than 50% activity of KpADH, indicating hydrophilic mutagenesis was not advantageous for the enantioselective binding and reduction of CPMK. Mutation of residues in PocC into aromatic amino acids resulted in less than 82% ee for all variants (Figure 1D), demonstrating that the decreased difference between size of PocC and PocC was unfavorable for the discrimination of 2′-pyridyl and 4′-chlorophenyl groups, which is in coincidence with the above hypothesis. Most variants of PocC-1 and PocC-3 were still active and exhibited higher ee, especially using Group I codons (Figure 1A, C). The three best variants in PocC-1, 3D3, 3H8, and 6C12, exhibited ee of 98.2%, 97.8%, and 98.5% respectively, significantly higher than that of KpADH (82.5% ee) (Table 2). Compared with PocC-2 and PocC-3, hydrophilic mutagenesis could alleviate the severe deactivation caused by hydrophilic mutagenesis to some extent. Consequently, hydrophobic interaction was important for the large substrate binding pocket. All the variants with over 95% ee in four HCSM libraries of PocC were rescreened and sequenced as shown in the Supporting Information. The mutational sites in 3D3, 3H8, and 6C12 are M131A/F197T/S237A, M131A/S237A, and M131V/F197T/S237A respectively.

Results of HCSM Libraries in PocC. Likewise, four HCSM libraries (PocC-1, PocC-2, PocC-3, and PocC-4) of C165S, E214Y, and Q238 were also developed and screened as shown in Figure 1E–H and Tables S3–S10. No severe deactivation was found in all four libraries. Variants with similar or higher specific activity toward CPMK than KpADH were selected for enantioselectivity analysis. Almost all the variants in PocC-1, PocC-2, and PocC-3 displayed lower ee than KpADH as illustrated in Figure 1E–G. Unlike the importance of hydrophobic interaction observed in PocC, steric hindrance plays a critical role in PocC. It is noted that mutation of C165S, E214Y, and Q238 into aromatic amino acids employing Group IV codons resulted in significantly improved ee (Figure 1H). The three best variants in PocC-IV, 43D11 (C165F/E214Y/Q238C), 50C10 (C165F/E214Y/S237A), and 51E10 (M131A/C165F/F197T/E214Y/S237A) exhibited ee of 97.4%, 97.5%, and 97.7%, respectively. As expected, mutation of residues in PocC into larger amino acids would decrease the size of PocC and further increase the discrepancy between PocC and PocC, which is favorable for the enantioselective recognition of prochiral ketones.

Results of Combinatorial Mutagenesis of PocC and PocC. Combinatorial mutageneses at hotspots of PocC and PocC were performed by mixing 3D3, 5H4, 6C12, 43D11, 45A3, and 47G9 as templates. About 250 variants in the combinatorial mutagenesis library were isolated and screened for improved activity and enantioselectivity. About 50% variants displayed improved activity as compared to KpADH and enantioselectivity of over 90% ee (Figure 1I and Table S11). The ee values of 49A7 (M131V/F197T/E214Y/S237A), 49D6 (C165F/E214Y/S237A/Q238C), 50C10 (C165F/E214Y/S237A), and 51E10 (M131A/C165F/F197T/E214Y/S237A) were 98.9%, 98.9%, 99.4%, and 98.7%, respectively, for all the variants isolated from PocC or PocC libraries. Remarkably, 50C10 exhibited the highest ee of 99.4% which satisfies the optical purity requirement of pharmaceutical intermediates. Although some variants also display high enantioselectivity, such as 50D8 (M131A/C165F/F197T/E214Y/S237A/Q238C) with 99.5% ee, they were eliminated due to lower specific activities than KpADH.

Kinetic Parameters and MD Simulation Analysis. To get insight into the effect of substrate binding pockets on the improved enantioselectivity, all the beneficial variants were purified to homogeneity, and their kinetic parameters were determined. Seven single mutants including M131V, C165S, F197T, E214Y, S237A, and Q238C were also constructed. The $k_{cat}/V_{max}$ and $k_{cat}/K_M$ of KpADH were 0.78 mM, 19.8 μmol-
min$^{-1}$mg$^{-1}$, and 16.9 s$^{-1}$·mM$^{-1}$, respectively. Variants of $P_{oc_1}$ displayed decreased $k_{cat}/K_M$ as compared with $Kp$ADH due to their higher $K_M$ (Table 2). Among M131V, F197T, and S237A, mutation at F197 exhibited a negative effect on substrate binding affinity, and the $K_M$ of F197T was as high as 8.12 mM, over 10 times that of $Kp$ADH. Similarly, the $K_M$ of 3D3 and 3H8 was also higher than that of $Kp$ADH. The $k_{cat}/K_M$ of 43D11, 45A3, and 47G9 were 10.4, 22.0, and 9.06 s$^{-1}$·mM$^{-1}$, respectively. The decreased $k_{cat}/K_M$ of 43D11 and 47G9 could be ascribed to the mutation of Q238C since the $K_M$ and $k_{cat}$ of C165F, E214Y, and Q238C were 0.52 mM and 9.26 s$^{-1}$, 0.80 mM and 14.8 s$^{-1}$, and 3.42 mM and 10.2 s$^{-1}$. A similar phenomenon was also found with 49A7, 49D6, and 51E10. Although the $k_{cat}$ values of 49A7, 49D6, and 51E10 were 28.4, 22.0, and 55.4 s$^{-1}$, much higher than 13.2 s$^{-1}$ of $Kp$ADH, the increased $K_M$ led to lower $k_{cat}/K_M$, considering all of them contain F197T and (or) Q238C. Both F197T and Q238C play important roles in influencing enantioselectivity; the ee values of 3D3 and 47G9 were slightly higher than those of 3H8 and 45A3 as shown in Table 2. The $k_{cat}/K_M$ of the best variant 50C10 (C165F/E214Y/S237A) was 59.3 s$^{-1}$·mM$^{-1}$, 3.51-fold that of $Kp$ADH, indicating the two-round screening was effective in identifying variants with high activity and enantioselectivity. The ee values of C165F, E214Y, and

Figure 2. Analysis of substrate binding pockets and interactions between ligand and protein in 20 ns MD simulation of $Kp$ADH and 50C10. (A) Substrate binding pocket of $Kp$ADH. (B) Substrate binding pocket of 50C10. (C) Interactions formed in the average conformation of $Kp$ADH–CPMK. (D) Interactions formed in the average conformation of 50C10–CPMK. CPMK is depicted in magenta, NADPH in yellow, catalytic triad in green, residues of binding pockets in cyan. Hydrogen bond is illustrated in green dotted line, $\pi$–$\pi$ stacking interaction in magenta line, halogen–alkyl interaction in pink line.

Figure 3. Substrate spectrum of $Kp$ADH and 50C10 toward various prochiral ketones.
S237A were 89.1%, 94.2%, and 96.1%, while their $k_{cat}/K_M$ values were 17.8, 18.6, and 40.5 s$^{-1}$mM$^{-1}$. The results indicate synergistic effects existed among C165, E214, and S237, since 50C10 displayed higher ee and $k_{cat}/K_M$ than three single mutants C165F, E214Y, and S237A.

To understand the molecular mechanism of improved enantioselectivity, the crystal structure of KpADH in complex with NADPH was resolved by X-ray diffraction and was deposited in the Protein Data Bank (PDB) under accession number 5Z2X (unpublished data). KpADH is a homodimer enzyme made up of 12 $\beta$-sheet surrounded by 10 $\alpha$-helix (Figure S2). The homology model of 50C10 was constructed by virtual mutation of KpADH. As illustrated in Figure 2, the substrate binding pocket of 50C10 is enlarged, and the surface becomes more hydrophobic than that of KpADH. Furthermore, substrate CPMK (1a) was docked into KpADH and 50C10, and the protein–ligand complexes were subjected to 20 ns molecular dynamic (MD) simulation (Figures S3–S4). The average conformations of KpADH–CPMK and 50C10–CPMK under catalytic state were extracted. The binding free energy between 50C10 and CPMK in the average conformation was calculated to be $-103.7$ kcal·mol$^{-1}$, which is $-31.2$ kcal·mol$^{-1}$ lower than that of KpADH–CPMK. The significantly decreased substrate binding energy could explain the decreased $K_M$ and increased $k_{cat}$ of 50C10. Interactions analysis reveals that only weak interactions could be formed between F197/V84 and CPMK in KpADH (Figure 2C). In addition, 50C10, more nonbonded interactions emerged, including halogen–alkyl interactions between M131/S237A and chlorine-substitute of CPMK and $\pi$–$\pi$ interactions between E214Y and pyridyl of CPMK as well as C165F and phenyl of CPMK (Figure 2D, Figure S5–S6). In addition, the increased difference between size of $P_{OC}$ and $P_{CC}$ is also important for the enantioselective recognition. The above analysis indicates that the stereoselectivity of the newly evolved 50C10 was enhanced by reshaping its substrate binding pocket employing HCSM strategy.

Substrate Spectrum of KpADH and 50C10. Various prochiral ketones with similar substituents as CPMK (1a) were chosen to evaluate the substrate specificity of KpADH and 50C10 as illustrated in Figure 3. For the reduction of chloroacetophenone derivatives (2a–5a), the specific activities and enantioselectivities of 50C10 were relatively lower than those of KpADH and in an order of 4′-chloroacetophenone $> 3′$-chloroacetophenone $> 2′$-chloroacetophenone. A similar phenomenon was also observed in the reduction of different acetylpyridines (6a–8a), except for 1-(pyridin-4′-yl)ethanone (8a). These results confirmed the importance of 2′-pyridyl and 4′-chlorophenyl groups in influencing the recognition and binding of 1a by KpADH. Eight diaryl ketones (9a–16a) were also selected to investigate the substrate profiles. Unlike chloroacetophenones or acetylpyridines, 50C10 displayed improved specific activities and enantioselectivities to almost all tested diaryl ketones. The specific activities of 50C10 toward 9a, 11a, 12a, 15a, and 16a were 1.02–7.54-fold that of KpADH. The ee values of 50C10 were significantly increased, especially toward 12a (97.5%, S), 13a (99.9%, S), 14a (99.9%, S), 15a (87.7%, S), and 16a (99.9%, S) with large electron-withdrawing substituents at para position, much higher than 12.3%–88.4% (S) of KpADH. Consequently, variant 50C10 could be potentially applied in the synthesis of optically pure diaryl alcohols with other electron-withdrawing substituents.

Asymmetric Preparation of Chiral (R)-1b Employing Variant 50C10. Considering its highest enantioselectivity and catalytic efficacy, 50C10 was applied in asymmetric reduction of CPMK (1a) for producing (R)-CPMA ((R)-1b) at high substrate loading (Figure 4). Glucose dehydrogenase from Bacillus megaterium (BmGDH) was introduced for cofactor regeneration. Tris-HCl (pH 7.5, 100 mM) was used as reaction buffer in which 50C10 and BmGDH retained relatively matched activities. As an initial attempt, 100 mM 1a (21.7 g·L$^{-1}$) with 10% ethanol as cosolvent was completely converted into (R)-1b within 2 h. Then, the substrate concentration was increased to 200 mM 1a (43.4 g·L$^{-1}$) with the same amount of enzymes. The conversion rate reached 72.1% at 2 h, and a full conversion was achieved at 6 h with 99.5% ee and no external NADP$^+$. Considering the superior enantioselectivity and substrate tolerance of 50C10 compared with other reported diaryl ketone reductases, much higher substrate loading was also investigated. Due to the low solubility of 1a, fed-batch of substrate was employed. As shown in Figure 4, 4.0, 4.0, and 2.0 mmol of 1a were added into the reaction mixture (a final concentration of 500 mM) at 0, 2, and 6 h. After 12 h of agitation, 1a was fully reduced with conversion of 99.7% and 99.4% ee. After extraction with ethyl acetate and evaporation under vacuum, about 1.94 g of white powder of (R)-1b was obtained with an isolation yield of 88.6%. The space–time yield reached 194 g·L$^{-1}$·d$^{-1}$ without assistance of external NADP$^+$.
CONCLUSIONS AND PERSPECTIVES

In summary, hydroclassified combinatorial saturation mutagenesis (HCSM) was proposed and applied in the directed evolution of enantioselectivity of alcohol dehydrogenase KpADH. Using this smart HCSM strategy, the library size could be significantly reduced especially with simultaneous mutation of three or more residues. The interactions including hydrophobicity and steric hindrance in the substrate binding pocket could be properly balanced employing the HCSM strategy by dividing 20 amino acids into four groups. Six hotspots critical for the enantioselectivity were identified in the substrate binding pockets of KpADH by alanine/tyrrosine scanning. After two rounds of screening, a highly enantioselective variant 50C10 was identified among 550 variants. 50C10 displayed high binding affinity and catalytic efficiency with $k_{cat}/K_m$ of 59.3 s$^{-1}$M$^{-1}$, 3.51-fold that of KpADH. Hydrophobic interaction in large substrate binding pocket and steric hindrance in small substrate binding pocket were proposed to be the main mechanisms for the improved enantioselectivity as proved by MD simulation. Substrate spectrum analysis revealed that 50C10 preferred diaryl ketones and displayed improved enantioselectivity, especially toward diaryl ketones with halogen or other electron-withdrawing substituents. Variant 50C10 also demonstrated high efficacy and substrate tolerance in the preparation of chiral diaryl alcohols and could convert as much as 0.5 M (108.5 g L$^{-1}$) (4-chlorophenyl)-(pyridin-2-yl)-methanone with 99.4% ee and space–time yield of 194 g L$^{-1}$ d$^{-1}$ without external NADPH$^\circ$. This newly proposed HCSM strategy could be potentially applied in the directed evolution of activity, substrate spectrum, and enantioselectivity especially when lacking of high throughput screening methods. This study also provides a highly efficient and enantioselective ketoreductase for the preparation of chiral diaryl alcohols.

MATERIALS AND METHODS

General Remarks. Plasmid pET28-kpadh was constructed in our previous work. Prochiral ketones, 4′-chlorophenyl)-(pyridin-2-yl) methanone (1a), 4′-chloroacetophenone (2a), 2′-chloroacetophenone (3a), 3′-chloroacetophenone (4a), 2′-chloro-1′-(4′-chlorophenyl)ethanone (5a), 1′-(pyridin-2′-yl)-ethanone (6a), 1′-(pyridin-3′-yl)ethanone (7a), 1′-(pyridin-4′-yl) ethanone (8a), benzophenone (9a), (pyridin-2′-yl)-phenylmethanone (10a), (4′-fluoroaryl)phenylmethanone (11a), (4′-chlorophenyl)phenylmethanone (12a), (4′-bromophenyl)phenylmethanone (13a), (4′-nitrophenyl)phenylmethanone (14a), (4′-methoxyphenyl)-phenylmethanone (15a), and 1,2-diphenylethanone (16a) were purchased from Aladdin Inc.. NADPH was bought from Merck Inc.

Site-Directed Mutagenesis and Protein Expression. Site-directed mutagenesis of alanine/tyrosine scanning was conducted by whole plasmid PCR using Phanta Super-Fidelity DNA Polymerase (Vazyme Inc.). Primers are listed in Table S1. PCR procedure was pre-denaturation at 96 °C for 5 min, 15 cycles of denaturation at 98 °C for 20 s, annealing at 50–65 °C for 20 s and elongation at 68 °C for 210 s, and final elongation at 68 °C for 10 min. The resultant PCR products were digested with DpnI to remove the parental pET28-kpadh. Furthermore, 5.0 μL of digested products were transformed into E. coli BL21(DE3). Expression of the KpADH variants was induced with 0.2 mM IPTG at 25 °C for 6 h as previously reported.

Activity Assay and Enantioselectivity Determination. The activity of KpADH variants was spectrophotometrically determined according to the optical changes of NADPH at 340 nm and 30 °C. Assay mixture contained 2.0 mM CPMK, 0.5 mM NADPH, and 10 μL enzyme in PBS buffer (100 mM, pH 6.0). One unit of activity was defined as the amount of enzyme required for the depletion of 1.0 μmol NADPH at the above-mentioned condition. Enantioselectivity was calculated according to the peak area of (R)- and (S)-CPMA determined by chiral HPLC equipped with Chiralcel OB-H column (4.6 mm × 250 mm × 5 μm, Daicel Chiral Technologies Co., Ltd.) as previously reported. All the assays were performed in triplicate.

Construction and Screening of HCSM Libraries. Overlap extension PCR was adopted to obtain the whole KpADH product with three mutational sites. Furthermore, the products were ligated into linearized pET28a by ClonExpress II (Vazyme Inc.). The resultant recombinant plasmids containing multiple mutations at M131/F197/S237 or C165/E214/Q238 were transformed into E. coli BL21(DE3). Different pairs of primers were used to form different HCSM libraries, and all the primers are listed in Table S2. About 600 monoclonals in each HCSM library were screened to achieve 95% coverage. Then all the strains were inoculated into 96-deep well plates and cultivated as above-mentioned for the expression of KpADH variants. After centrifugation, cells were freeze-thawed at −80 °C followed by lysozyme and DNaseI treatment to obtain crude extracts. Two rounds of screening were adopted. First, the activities of all the variants were high throughout determined as above-mentioned. Then, the enantioselectivities of variants with similar or higher activity than KpADH were determined.

Protein Purification and Preparation of Crude Enzyme Powder. KpADH and variants with His-tag were purified to homogeneity by nickel-affinity chromatography as previously reported and analyzed by SDS-PAGE. Fermentation of 50C10 was carried out with 2 × LB as fermentation medium. Cells harboring 50C10 were harvested, washed with physiological saline, and disrupted with nanohomogenize machine (ATS AH-BASIC1). The crude extract was obtained by centrifugation and lyopholized under vacuum (SCIENTZ-10N) to form the crude enzyme powder, which was stored at 4 °C for further use.

Determination of Kinetic Parameters and Substrate Profiles. Kinetic parameters of KpADH and variants were determined. The specific activities of each variant toward 0.1–5.0 mM 1a were assayed employing the above-mentioned protocol. $K_M$, $V_{max}$ and $k_{cat}$ were calculated according to the Lineweaver–Burk plot. Substrate profiles of KpADH and 50C10 were measured toward prochiral ketones 1a–16a. The final substrate concentration of each substrates was 2.0 mM. The analytical methods for enantioselectivity of all tested substrates are summarized in Table S12. All the activities and enantioselectivities were measured in triplicate.

Asymmetric Preparation of Chiral (R)-CPMA. Asymmetric reduction of 1a for the preparation of (R)-CPMA was performed with 0.30 and 0.10 g of crude enzyme powder of 50C10 and GDH, 2.0–10.0 mmol 1a dissolved in 2 mL of ethanol, and 3.0–7.5 mmol glucose in 18 mL of Tris-HCl (pH 7.5, 100 mM). Reactions were magnetically agitated at 200 rpm and 30 °C and maintained at pH 7.5 using 1.0 M NaOH. Samples were withdrawn from the reaction mixture at different time points, and the conversion and enantioselectivity were analyzed as above-mentioned. After the reaction, the mixture

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was extracted three times with ethyl acetate. The organic phase was combined and dried over anhydrous Na₂SO₄. Furthermore, the organic phase was evaporated under vacuum to obtain (R)-1b as white powder.

**Homology Modeling and Molecular Docking Analysis.** Protein modeling was implemented with EasyModeller 4.0 using protein structures 4PVC and 5B6K as templates and verified with SAVES (Structure Analysis and Verification Server version 4). All docking calculations were accomplished with AutoDock Vina 1.1, with the docking algorithm that took account of ligand flexibility but kept the protein rigid. Docking runs were carried out using standard parameters of the program for interactive growing and subsequent scoring. Molecular dynamics simulation was carried out employing the NAMD module of Discover Studio 3.5. CHARMM force field and temperature of 300 K were adopted, and the simulation time was set as 20 ns. Stereoviews were constructed using Pymol.

**ASSOCIATED CONTENT**

1. Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscatal.8b02286.

Supporting tables and figures for primers, creation and screening of HCSM libraries, enantioselectivity analysis, original data of kinetic parameters and substrate profiles, 

1H NMR and 13C NMR of synthesized (R)-CPMA, and crystal structure of KpADH (5Z2X) (PDF)

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**Notes**

The authors declare no competing financial interest.

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